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mass unit (u): "One twelfth of the rest mass of an unbound carbon-12, in its nuclear and electronic ground state, which is equal to  $1.661 \times 10^{-27}$  kg." Protons, neutrons and electrons all taken into account.

Atomic mass: "The average mass of an atom in u." Shown on the periodic table. Mole (n): "The amount of substance having the same number of

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